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Protecting-Group-Free Total Synthesis of (−)-Jiadifenolide: Development of a $[4 + 1]$ Annulation toward Multisubstituted Tetrahydrofurans

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S Supporting Information

[AB](#page-3-0)STRACT: [A concise, pr](#page-3-0)otecting-group-free total synthesis of (−)-jiadifenolide, a synthetically challenging seco-prezizaane sesquiterpene with potent neurotrophic activity, is reported. The convergent route features a $SmI₂/H₂O$ -mediated stereoselective reductive cyclization, an unprecedented formal [4 + 1] annulative tetrahydrofuran-forming reaction and programmed redox manipulations. The newly developed annulation of β -hydroxy aldehydes or ketones with lithium trimethylsilyldiazomethane provides access to a diverse array

of multisubstituted tetrahydrofurans. The synthetic jiadifenolide exhibited weak cytotoxicity against five human cancer cell lines.

Terpenes exhibit a broad range of biological activities and have been widely used in medicine and other industries.¹ Chemically, they often provide opportunities for the discovery and invention of new ring-forming reactions in organi[c](#page-3-0) synthesis.² The genus Illicium is a rich source for chemically and biologically intriguing secondary metabolites, especially sesquiter[pe](#page-3-0)nes.³ Many of them have attracted considerable attention from chemists and biologists.⁴

(−)-Jiadifen[ol](#page-3-0)ide (1, Scheme 1) is a complex sesquiterpene isolated in 0.0009% yield from the dr[ie](#page-3-0)d pericarps of Illicium jiadifengpi by Fukuyama and co-workers in 2009 .⁵ It displayed prominent neurotrophic activity at low concentration (10 nM), which renders it a promising lead compound f[or](#page-3-0) developing therapeutics for neurodegenerative diseases.⁶ Structurally, jiadifenolide features a highly oxygenated cagelike architecture

with seven contiguous stereocenters, in particular, a highly congested cyclohexane (B ring) containing four quaternary stereocenters. Notably, the diverse oxygenation patterns of 1, such as α -hydroxy lactone and α -hemiketal lactone, posed considerable challenges for the differentiation of functional groups in chemical synthesis, and thus, protecting-group manipulations seemed inevitable. Its fascinating structure and potential therapeutic value have attracted extensive synthetic interest, thus resulting in four elegant total syntheses of jiadifenolide by Theodorakis (25 steps) , Sorensen (18 steps) ,⁸ and Paterson $(23 \text{ steps})^9$ and very recently by Shenvi (eight steps).¹⁰ Although preliminary evalu[ati](#page-3-0)ons of neurotrophi[c](#page-3-0) activity have been carrie[d](#page-3-0) out by the groups of Fukuyama⁵ and Theo[dor](#page-3-0)akis, 11 full biological profiling of jiadifenolide, which is indispensable for its development toward a therapeutic ag[en](#page-3-0)t, is still pending. [H](#page-3-0)erein, we describe a concise, protecting-groupfree total synthesis¹² of $(-)$ -jiadifenolide. Furthermore, we also present a case study in which the total synthesis of complex natural products d[ire](#page-3-0)ctly drives the discovery and development of a new formal $[4 + 1]$ annulation toward tetrahydrofurans.

Cyclization and oxidation are extensively studied topics in terpenoid synthesis, 13 and the deliberate arrangement of these two kinds of manipulations could dramatically enhance the efficiency of synthe[sis](#page-3-0). As outlined in Scheme 1, we envisaged that the pentacyclic skeleton of 1 could be assembled via three diastereoselctive cyclizations from the known β -keto ester 5 :¹⁴ a sequence of oxidations followed by hemiketalization to form

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the D ring $(2 \rightarrow 1)$, formal $[4 + 1]$ annulation to generate the tetrahydrofuran E ring $(3 \rightarrow 2)$, and reductive cyclization mediated by samarium diiodide and water to create the central B ring $(4 \rightarrow 3)$. We also anticipated that programmed redox manipulations would install the requisite oxygen-containing functional groups and enable an "economic"¹⁵ and protectinggroup-free total synthesis of sesquiterpene 1.

As depicted in Scheme 2, our synthesis [co](#page-3-0)mmenced with diastereoselective allylation of the known β -keto ester 5, which

could be prepared in large scale from $(R)-(+)$ -pulegone in a modified two-step procedure (see the Supporting Information). Subsequent ozonolytic cleavage of the resulting terminal double bond afforded aldehyde 6 in 91% yield. Selective aldol addition between the aldehyde group of 6 and a boron furanolate¹⁶ generated in situ from butenolide 7 provided Baylis−Hillmantype adducts 8 as a mixture of inseparable diastereomers, whi[ch](#page-3-0) underwent dehydration via an acetylation−elimination cascade to deliver diene 9 with an overall yield of 71%. To circumvent the utilization of protecting groups and achieve chemoselectivities between the oxygenated functional groups, we designed a group of programmed reduction manipulations. First, a selective reduction of the hindered ester group of 9 in the presence of a ketone and an unsaturated lactone functions was achieved by in situ conversions of the ketone and lactone to enolates 17 by treatment with LDA prior to the introduction of DIBAL-H. Subsequently, a hydroxy-directed selective hydrogena[tio](#page-3-0)n¹⁸ of disubstituted double bond (C7−C8) in the resulted dienol was achieved by Adam's catalyst under a H_2 atmosphere (1 atm) ,¹⁹ providing 4 in 85% yield.

At this stage, we turned our attention to the construction of the central, highly [su](#page-3-0)bstituted B ring via a $SmI₂$ -mediated reductive cyclization.²⁰ In Paterson's pioneering work,⁹ a similar strategy was also conceived to close the B ring. However, due to low diastereoselectiv[ity](#page-3-0) in the aldol reaction and mod[er](#page-3-0)ate yield of the SmI₂-mediated reductive cyclization, the overall efficiency of this key transformation was not satisfactory. After extensive studies of this key transformation over a series of substrates (for details, see Supporting Information), we

finally chose compound 5 as the cyclization precursor. Our initial attempts with traditional conditions (such as SmI₂ in THF, $SmI_2/LiCl/t-BuOH$ in THF) only resulted in ketonereduction products. Enlightened by Flowers' ²¹ and Procter's findings,²² we decided to use H_2O to modulate the redox potential of $SmI₂$ and stabilize the possible rad[ica](#page-3-0)l intermediate. To our [de](#page-3-0)light, with a large amount of H_2O (7000 equiv) as cosolvent, treatment of 4 with 2.2 equiv of $SmI₂$ provided the desired tricycle 10 in 70% yield. The relative stereochemistry of diol 10 was confirmed by X-ray crystallography. The stereochemical outcome of this transformation can be rationalized as the result of ketyl radical addition to unsaturated lactone via a chelated chair/twist boat conformation shown as 4TS (Scheme 2). Subsequent Swern oxidation smoothly delivered β -hydroxy aldehyde 3 in quantitative yield.

Serendipitous discoveries have led to many new reactions and thus inspired new approaches to natural products.²³ Initially, we planned to access the E ring via a terminal alkyne, which was supposed to be obtained from aldehyde 3 [by](#page-3-0) treatment with lithium trimethylsilyldiazomethane (TMSC- $(Li)N_2$).²⁴ However, a novel formal $[4 + 1]$ annulative tetrahydrofuran-forming reaction²⁵ was discovered that dramaticall[y e](#page-3-0)xpedited the synthetic journey (see the Supporting Information for the condition o[ptim](#page-3-0)ization and see below for further investigation of this novel reaction). Specifically, with LiCl as an additive, reaction of β -hydroxy aldehyde 3 with TMSC(Li)N₂ at -78 °C afforded 3-trimethylsiloxy furanoid 2 via a cascade process in 78% yield as a single diastereomer. It should be noted that the reaction temperature was critical. When the reaction mixture was warmed to 0° C before being quenched with saturated $NAHCO₃$ solution, the only identified product was dihydrofuran species (see the Supporting Information). The configuration of the newly built stereocenter (C12) in 2 was assigned by NOE NMR spectroscopic studies and further confirmed by the X-ray crystal structure of a later intermediate 11 (Scheme 3).

With tetracycle 2 in hand, our efforts were then focused on the incorporation of oxygen-containing functions. A program of oxidation manipulations finalized the actual oxidation state of jiadifenolide (Scheme 3). Dehydrogenation at C6−C7 via a

one-pot phenylselenation and oxidative elimination furnished unsaturated lactone 11, which was further converted to epoxide 12 with DMDO. 3-Trimethylsiloxy furanoid 12 was oxidized under Sharpless conditions, 26 probably via a 3-furanone intermediate to the α -keto lactone 13, which possesses the same oxidation level as that [of t](#page-3-0)he natural product.

Finally, a cascade reaction triggered by LiOH^{8,27} afforded (−)-jiadifenolide (1) in 71% yield, and more than 300 mg of the final product was obtained. The synthetic sam[ple](#page-3-0) exhibited spectroscopic properties identical with those reported for the natural product. $5,28$ The absolute stereochemistry of 1 was confirmed by an X-ray crystal structure in conjunction with the absolute stereoc[hem](#page-3-0)istry of C1 inherited from $(R)-(+)$ -pulegone.

Encouraged by the success of the new formal $[4 + 1]$ annulation in the total synthesis, we next investigated the generality of this reaction (Scheme 4). By treatment with

^aThe relative stereochemistry of the products was not determined. but continue the relationship of the products was determined by NOE NMR spectroscopic studies and the analogy.

TMSC(Li)N₂ at -78 °C, a collection of β -hydroxy aldehydes (14a−f) were smoothly converted to the corresponding tetrahydrofuran derivatives (15a−f) in moderate to high yield, featuring monocyclic, spirocyclic, and fused ring systems. To our delight, this method was also viable to more challenging $β$ -hydroxy ketone substrates, and a set of $β$ -hydroxy ketones (14g−j) was transformed to complex furanoids (15g−j) in moderate to good yield. To our surprise, when β -hydroxy

ketone 14h was treated by $TMSC(Li)N_{2}$, not only the cis-fused tetrahydrofuran (cis-15h) but also the unusual trans-fused²⁹ diastereomer with high ring string (trans-15h) was obtained (29% yield, $cis/trans = 1:1.2$). Thus, we provided a hig[hly](#page-3-0) effective approach for the construction of multisubstituted tetrahydrofurans from readily available β-hydroxy aldehyde or ketone. Furthermore, current transformation could also serve as a stepstone for the preparation of α -keto lactones (e.g., 12 \rightarrow 13).

As outlined in Scheme 5, a possible mechanism for the novel formal $[4 + 1]$ annulation toward tetrahydrofurans is proposed.

Scheme 5. Proposed Mechanism for the Formal $[4 + 1]$ Annulation

The nucleophilic addition of $TMSC(Li)N₂$ to the aldehyde/ ketone 14 with the help of lithium ion complexation afforded adduct II, which usually decomposed to a terminal alkyne when elevating the reaction temperature.²⁴ However, a tandem Brook rearrangement³⁰ and 1,5-proton transfer occurred to give intermediate III. Decomposition [of](#page-3-0) diazo compound III to carbene IV f[oll](#page-3-0)owed by addition of the alkoxide anion to carbene 31 afforded anion V, which was converted to trimethylsiloxy furanoid 15 by protonation (pathway a). We also ca[nno](#page-3-0)t completely exclude that 15 could be partially obtained via a protonation and intramolecular carbene O−H insertion³² sequence (pathway b). As a whole, the tetrahydrofuran was formed through a formal $[4 + 1]$ annulation between [an](#page-3-0) amphiphilic β-hydroxy aldehyde or ketone and a carbene precursor.

As a part of our project of biological evaluation of jiadifenolide and its analogues, the cytotoxicity of the synthetic jiadifenolide against five human cancer cell lines was evaluated. It was found that 1 only showed weak antitumor activities (IC^{50}) $> 100 \mu M$) (see the Supporting Information).

In summary, a concise, protecting-group-free total synthesis of the complex neurotrophic sesquiterpene (−)-jiadifenolide has been accomplished in 13 steps with an overall yield of 7.9% from known compound 5 (15 steps from $(R)-(+)$ -pulegone). A newly developed formal $[4 + 1]$ annulative tetrahydrofuranforming reaction guaranteed the synthetic efficiency. Furthermore, this novel method exhibited good feasibility for accessing a diverse array of multisubstituted tetrahydrofurans. We expect the reaction will find more use in the synthesis of natural products and pharmaceuticals. Further exploration of the novel annulation and biological evaluation of (−)-jiadifenolide are under way in our laboratory.

Organic Letters
■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02845.

> Experimental details and spectral data for all new compounds $(^1H$ NMR, ^{13}C NMR, IR, and HRMS) (PDF)

X-ray data for compounds 10, 11 and 1 (CIF)

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Notes

The authors declare no competing financial interest.

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■ **DEDICATION**

This work is dedicated to Prof. Zhen Yang (Peking University).

■ REFERENCES

(1) (a) Dewick, P. M. Medicinal Natural Products; Wiley: Hoboken, 2009; pp 187−310. (b) Breitmaier, E. Terpenes; Wiley-VCH: Weinheim, 2006.

(2) (a) Jansen, D. J.; Shenvi, R. A. Future Med. Chem. 2014, 6, 1127. (b) Maimone, T. J.; Baran, P. S. Nat. Chem. Biol. 2007, 3, 396.

(3) (a) Liu, Y.-N.; Su, X.-H.; Huo, C.-H.; Zhang, X.-P.; Shi, Q.-W.; Gu, Y.-C. Chem. Biodiversity 2009, 6, 963. (b) Fukuyama, Y.; Huang, J.- M. Chemical Constituents of the Genus Illicium. In Illicium, Pimpinella and Foeniculum; Jodral, M. M., Ed.; CRC: Boca Raton, FL, 2004; pp 31−68.

(4) For a review, see: (a) Urabe, D.; Inoue, M. Tetrahedron 2009, 65, 6271. For recent works, see: (b) Yang, Y.; Fu, X.; Chen, J.; Zhai, H. Angew. Chem., Int. Ed. 2012, 51, 9825. (c) Chen, J.; Gao, P.; Yu, F.; Yang, Y.; Zhu, S.; Zhai, H. Angew. Chem., Int. Ed. 2012, 51, 5897. (d) Ogura, A.; Yamada, K.; Yokoshima, S.; Fukuyama, T. Org. Lett. 2012, 14, 1632. (e) Trzoss, L.; Xu, J.; Lacoske, M. H.; Mobley, W. C.; Theodorakis, E. A. Org. Lett. 2011, 13, 4554. (f) Shi, L.; Meyer, K.; Greaney, M. F. Angew. Chem., Int. Ed. 2010, 49, 9250.

(5) Kubo, M.; Okada, C.; Huang, J.-M.; Harada, K.; Hioki, H.; Fukuyama, Y. Org. Lett. 2009, 11, 5190.

(6) Wilson, R. M.; Danishefsky, S. J. Acc. Chem. Res. 2006, 39, 539. (7) Xu, J.; Trzoss, L.; Chang, W. K.; Theodorakis, E. A. Angew. Chem., Int. Ed. 2011, 50, 3672.

(8) Siler, D. A.; Mighion, J. D.; Sorensen, E. J. Angew. Chem., Int. Ed. 2014, 53, 5332.

(9) Paterson, I.; Xuan, M. Y.; Dalby. Angew. Chem., Int. Ed. 2014, 53, 7286.

(10) Lu, H.-H.; Martinez, M. D.; Shenvi, R. A. Nat. Chem. 2015, 7, 604.

(11) Trzoss, L.; Xu, J.; Lacoske, M. H.; Mobley, W. C.; Theodorakis, E. A. Chem. - Eur. J. 2013, 19, 6398.

(12) For reviews, see: (a) Young, I. S.; Baran, P. S. Nat. Chem. 2009, 1, 193. (b) Hoffmann, R. W. Synthesis 2006, 2006, 3531.

(13) (a) Chen, K.; Baran, P. S. Nature 2009, 459, 824. (b) Hudlicky, T.; Reed, J. W. The Way of Synthesis; Wiley-VCH: Weinheim, 2007.

(14) (a) Miyagawa, T.; Nagai, K.; Yamada, A.; Sugihara, Y.; Fukuda,

T.; Fukuda, T.; Uchida, R.; Tomoda, H.; Ōmura, S.; Nagamitsu, T.

Org. Lett. 2011, 13, 1158. (b) Marx, J. N.; Norman, L. R. J. Org. Chem. 1975, 40, 1602.

(15) (a) Burns, N. Z.; Baran, P. S.; Hoffmann, R. W. Angew. Chem., Int. Ed. 2009, 48, 2854. (b) Wender, P. A.; Verma, V. A.; Paxton, T. J.; Pillow, T. H. Acc. Chem. Res. 2008, 41, 40. (c) Trost, B. M. Science 1991, 254, 1471.

(16) (a) Boukouvalas, J.; Wang, J.-X.; Marion, O. Tetrahedron Lett. 2007, 48, 7747. (b) Jefford, C. W.; Jaggi, D.; Boukouvalas, J. J. Chem. Soc., Chem. Commun. 1988, 1595.

(17) (a) Devineau, A.; Grosbois, M.; Carletti, I.; Vacher, B. J. Org. Chem. 2009, 74, 757. (b) Takikawa, H.; Tobe, M.; Isono, K.; Sasaki, M. Tetrahedron 2005, 61, 8830. (c) Wright, J.; Drtina, G. J.; Roberts, R. A.; Paquette, L. A. J. Am. Chem. Soc. 1988, 110, 5806.

(18) For a review, see: Hoveyda, A. H.; Evans, D. A.; Fu, G. C. Chem. Rev. 1993, 93, 1307.

(19) When the ¹H NMR spectroscopic analysis of the reaction mixture indicated no starting material remained, the reaction was stopped.

(20) For reviews, see: (a) Szostak, M.; Fazakerley, N. J.; Parmar, D.; Procter, D. J. Chem. Rev. 2014, 114, 5959. (b) Edmonds, D. J.; Johnston, D.; Procter, D. J. Chem. Rev. 2004, 104, 3371.

(21) (a) Prasad, E.; Flowers, R. A. J. Am. Chem. Soc. 2005, 127, 18093. (b) Chopade, P. R.; Prasad, E.; Flowers, R. A. J. Am. Chem. Soc. 2004, 126, 44.

(22) (a) Szostak, M.; Spain, M.; Procter, D. J. J. Am. Chem. Soc. 2014, 136, 8459. (b) Parmar, D.; Price, K.; Spain, M.; Matsubara, H.; Bradley, P. A.; Procter, D. J. J. Am. Chem. Soc. 2011, 133, 2418.

(23) (a) Martin, D. B. C.; Vanderwal, C. D. A Short Synthesis of Strychnine from Pyridine. In Total Synthesis of Natural Products; Li, J. J., Corey, E. J., Eds.; Springer: Berlin, 2012; pp 67−102. (b) McNally, A.; Prier, C. K.; MacMillan, D. W. C. Science 2011, 334, 1114. (c) Roberts, R. M. Serendipity: Accidental Discoveries in Science; Wiley: New York, 1989.

(24) Colvin, E. W.; Hamill, B. J. J. Chem. Soc., Chem. Commun. 1973, 151.

(25) A similar conversion was observed: (a) Higuchi, R.; Tokimitsu, Y.; Komori, T. Liebigs Ann. Chem. 1988, 1988, 249. For recent developments, see: (b) Nicolle, S. M.; Lewis, W.; Hayes, C. J.; Moody, C. J. Angew. Chem., Int. Ed. 2015, 54, 8485. (c) Lindsay, V. N. G.; Viart, H. M.-F.; Sarpong, R. J. Am. Chem. Soc. 2015, 137, 8368.

(26) Carlsen, P. H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936.

(27) (a) Sumiya, T.; Ishigami, K.; Watanabe, H. Angew. Chem., Int. Ed. 2010, 49, 5527. (b) Fawcett, J.; Griffith, G. A.; Percy, J. M.; Uneyama, E. Org. Lett. 2004, 6, 1277.

(28) The ¹H NMR, ¹³C NMR, and IR spectra of synthetic 1 matched those previously reported by Fukuyama in ref 5. For details, see the Supporting Information.

(29) For the selected examples for the synthesis of trans 5/5 fused rings, see: (a) Pronin, S. V.; Shenvi, R. A. Nat. Chem. 2012, 4, 915. (b) Jamison, T. F.; Shambayati, S.; Crowe, W. E.; Schreiber, S. L. J. Am. Chem. Soc. 1994, 116, 5505.

(30) (a) Vitale, M.; Lecourt, T.; Sheldon, C. G.; Aggarwal, V. K. J. Am. Chem. Soc. 2006, 128, 2524. (b) Aggarwal, V. K.; Sheldon, C. G.; Macdonald, G. J.; Martin, W. P. J. Am. Chem. Soc. 2002, 124, 10300.

(31) For nucleophilic addition to carbene species, see: (a) Yanagisawa, H.; Miura, K.; Kitamura, M.; Narasaka, K.; Ando, K. Helv. Chim. Acta 2002, 85, 3130. (b) Topolski, M. J. Org. Chem. 1995, 60, 5588. (32) For a review, see: Miller, D. J.; Moody, C. J. Tetrahedron 1995, 51, 10811.